

## Novel Asymmetric Synthesis of Penaresidin B as a Potent Actomyosin ATPase Activator

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**Abstract:** An efficient and novel synthetic process is described for the preparation of an azetidine ring with the contiguous stereogenic centers and the total synthesis of penaresidin B by featuring the elaboration of the functionalized homochiral lactam derived from D-glutamic acid.  
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Penaresidin A (**1**) and B (**2**), first isolated in 1991 from an Okinawan marine sponge *Penares* sp. by Kobayashi et al.<sup>1</sup>, exhibit potent actomyosin ATPase-activating activity. After structural characterization as a mixture of the corresponding tetraacetyl derivatives, these were revealed to be the first sphingosine-derived alkaloids<sup>2</sup> possessing an azetidine ring. During our synthetic studies of these substances, the syntheses of both a straight-chain analog by Kamikawa et al.<sup>3</sup> and three stereoisomers of **1** with *2S,3R,4S*-configurations of the azetidine ring and syn configuration between C-15 and C-16 of the side chain by Mori et al.<sup>4</sup> have been established. In addition, recently the initially proposed structure of penaresidin B was revised to be **2**<sup>5</sup> as shown in Fig. 1 and the absolute configuration at C-15 in **1** and **2** has been determined to be *S*.<sup>6</sup>

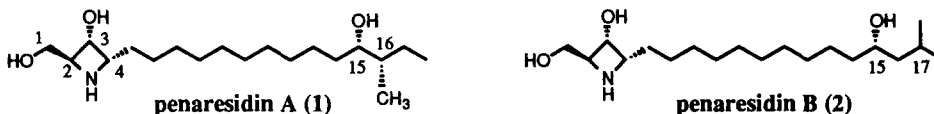
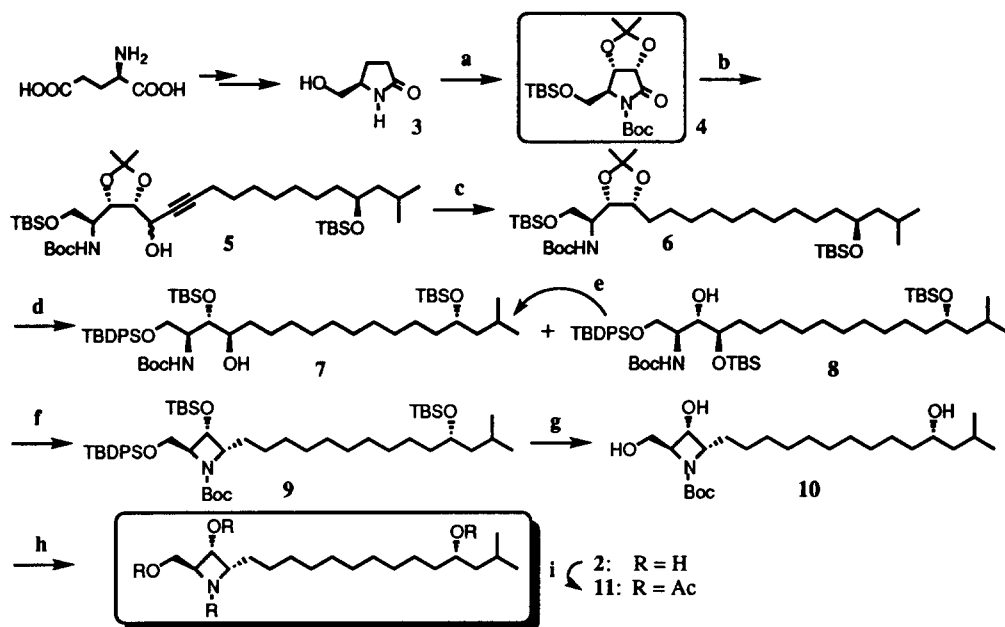


Fig. 1

With the above stereochemical outcome in mind, the central feature of this report is to communicate the details of the novel and expeditious route from D-glutamic acid for the construction of the azetidine ring with the desired contiguous stereogenic centers and the total synthesis of penaresidin B (**2**).

*N*-Boc lactam **4** obtained from D-glutamic acid through lactam **3** (99% ee) was treated with the acetylide elaborated from D-leucine in 9 steps<sup>7</sup> via the acetylene zipper reaction,<sup>8</sup> followed by the reduction with NaBH<sub>4</sub> of the corresponding tautomer of ketoamide and hydroxy pyrrolidine form<sup>9</sup> to give the alcohol **5** as a diastereomer mixture. Formation of thioimidazolidine and successive radical deoxygenation with Bu<sub>3</sub>SnH<sup>10</sup> resulted in the clean preparation of **6**, [α]<sub>D</sub><sup>18</sup>+27.3° (c 1.47, CHCl<sub>3</sub>). After deprotection of the four hydroxyl groups of **6**, reprotection of the primary alcohol with TBDPSCl and the secondary ones with TBSCl brought about undesirable regioselectivity, leading to the mixture of alcohols **7** and **8** (2:98 determined by HPLC) in high yield. Fortunately, it has become apparent that the major product **8** was smoothly transformed into the desired **7** with 1,4-silyl rearrangement under basic conditions.<sup>11</sup> After chromatographic isolation of **7**, [α]<sub>D</sub><sup>20</sup>+10.4° (c 1.69, CHCl<sub>3</sub>), mesylation and cyclization with NaH were subsequently submitted to produce **9**, [α]<sub>D</sub><sup>24.5</sup>+40.0° (c 0.36, CHCl<sub>3</sub>) with the azetidine ring containing the same contiguous configurations as those of natural one. Finally, **9** was treated with Bu<sub>4</sub>NF to give the *N*-Boc derivative **10**, [α]<sub>D</sub><sup>20</sup>+42.6° (c 0.23,



**Scheme 1.** Reagents and conditions:

(a) 6 steps; see ref. 2; (b) 1, (*S*)-HC≡C(CH<sub>2</sub>)<sub>7</sub>CH(OTBS)CH<sub>2</sub>CH(CH<sub>3</sub>)<sub>2</sub>, BuLi, THF, -78 °C; 2, NaBH<sub>4</sub>, EtOH; 58% (2 steps); (c) 1, (thiocarbonyl)diimidazole, THF, 40 °C; 98%; 2, Bu<sub>3</sub>SnH, AIBN, toluene, 95 °C; 70%; (d) 1, *p*-TsOH, MeOH; 63%; 2, TBDPSCl, Et<sub>3</sub>N, CH<sub>2</sub>Cl<sub>2</sub>; 75%; 3, TBSCl, imidazole, DMF; 98% (8 : 9 = 2 : 98); (e) NaH, THF, 35% (conv. 80%); (f) 1, MsCl, pyridine; 2, NaH, THF, 58% (2 steps); (g) Bu<sub>4</sub>NF, THF, 79%; (h) 1, BF<sub>3</sub>·OEt<sub>2</sub>, CH<sub>2</sub>Cl<sub>2</sub>, -40~-30 °C; (i) Ac<sub>2</sub>O, pyridine, CH<sub>2</sub>Cl<sub>2</sub>; quant. (2 steps).

MeOH) of penaresidin B (2) and the *N*-Boc group was removed with BF<sub>3</sub>·OEt<sub>2</sub> to complete the synthesis of 2. The structure was confirmed by conversion to the known tetraacetate 11, [ $\alpha$ ]<sub>D</sub><sup>25</sup>+49.1° (c 0.695, CHCl<sub>3</sub>) [lit. [ $\alpha$ ]<sub>D</sub><sup>25</sup>+47° (c 0.42, CHCl<sub>3</sub>)],<sup>5</sup> whose physical data were completely identical with the reported values in all respects.<sup>5</sup>

This process starting from D-glutamic acid provides a new synthetic strategy and represents a short and easily accessible pathway to penaresidins.

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#### References and notes

- Kobayashi, J.; Cheng, J.-f.; Ishibashi, M.; Wälchli, M. R.; Yamamura, S.; Ohizumi, Y. *J. Chem. Soc., Perkin Trans. 1* **1991**, 1135-1137.
- Yoda, H.; Oguchi, T.; Takabe, K. *Tetrahedron: Asymmetry* **1996**, *7*, 2113-2116.
- Hiraki, T.; Yamagiwa, Y.; Kamikawa, T. *Tetrahedron Lett.* **1995**, *36*, 4841-4844.
- Takikawa, H.; Maeda, T.; Mori, K. *Tetrahedron Lett.* **1995**, *36*, 7689-7692.
- Takikawa, H.; Maeda, T.; Seki, M.; Koshino, H.; Mori, K. *J. Chem. Soc., Perkin Trans. 1* **1997**, 97-111.
- Kobayashi, J.; Tsuda, M.; Cheng, J.-f.; Ishibashi, M.; Takikawa, H.; Mori, K. *Tetrahedron Lett.* **1996**, *37*, 6775-6776.
- Mori, K. *Tetrahedron* **1976**, *32*, 1101-1106.
- a) Brown, C. A.; Yamashita, A. *J. Am. Chem. Soc.* **1975**, *97*, 891-892.  
b) Abrams, S. R. *Can. J. Chem.* **1984**, *62*, 1333-1334.
- Yoda, H.; Yamazaki, H.; Kawauchi, M.; Takabe, K. *Tetrahedron: Asymmetry* **1995**, *6*, 2669-2672.  
b) Yoda, H.; Nakajima, T.; Takabe, K. *Tetrahedron Lett.* **1996**, *37*, 5531-5534.
- Rasmussen, J. R.; Slinger, C. J.; Kordish, R. J.; Newman-Evans, D. D. *J. Org. Chem.* **1981**, *46*, 4843-4846.
- After detailed investigations, the best result was observed under the conditions employing NaH as a base in THF.